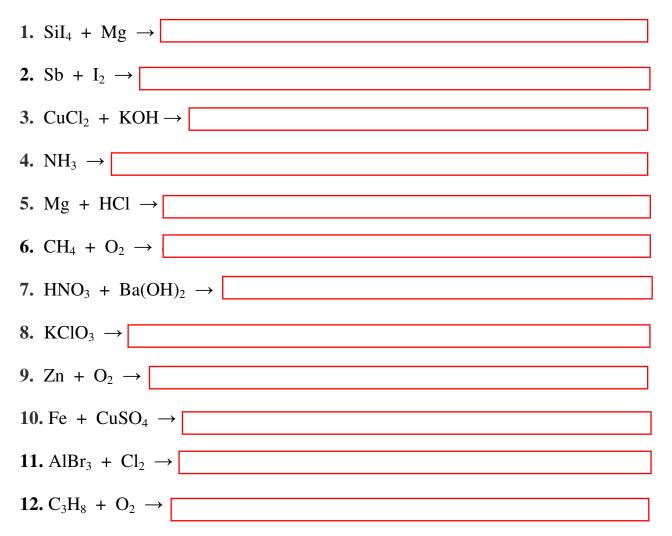
Write the chemical formulas for the products in each equation. Make sure that the charge on each ionic compound is neutral.



## Using the information below write complete equations for the following reactions. Balance each equation.

- 13. Lithium metal reacts with liquid bromine.
- 14. Potassium metal reacts with silver chloride.
- **15.** When sodium metal reacts with hydrochloric acid bubbles of a combustible gas form.